

Citrate-based biodegradable injectable hydrogel composites for orthopedic applications

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Previous studies have confirmed that natural bone apatite crystals are bound with citrate-rich molecules. Citrates on apatite crystals impact bone development and its load-bearing function. However, such understanding has never been translated into bone biomaterials design. Herein, the first citrate-based injectable composite material for orthopedic applications is developed based on our recently developed biodegradable poly(ethylene glycol) maleate citrate (PEGMC) and hydroxyapatite (HA). PEGMC contains rich carboxylic groups that could chelate with calcium-containing HA, thus facilitating polymer/HA interactions, similar to natural citrate-bound apatite crystal. The crosslinking of poly(ethylene glycol) diacrylate (PEGDA) with PEGMC/HA composites allows additional control over the degradation and mechanical properties of the crosslinked PEGMC/HA (CPEGMC/HA) composites. CPEGMC/HA composite can serve as an ideal injectable cell carrier as confirmed by the enhanced DNA content, ALP activity, and calcium production through a human fetal osteoblast encapsulation study. An *ex vivo* study on a porcine femoral head demonstrated that PEGMC/HA is a potentially promising injectable biodegradable bone material for the treatment of osteonecrosis of the femoral head. Development of biodegradable citrate-based injectable PEGMC/HA composite is an initial step for the development of the next generation of bone tissue engineering and orthopedic biomaterials.

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Introduction

Mimicking nature with respect to the tissue molecular, biological, and dynamic environment, and the tissue's compositional and morphological structure has been a widely adopted strategy and the ultimate goal in tissue engineering scaffolding design. Significant progress in bioconjugation,^{1–3} growth factor delivery,⁴ nanostructure fabrication,⁵ recombinant proteins/matrices,^{6–8} self-assembly peptides,^{9,10} and stem cell niches^{11,12} has been made and are all within the scope of mimicking nature in tissue engineering. Bone scaffolding materials have been heavily studied in the past two decades, probably the most studied in tissue engineering. Examples include natural polymers such as polysaccharides (starch, alginate, chitosan, hyaluronic acid) and synthetic biodegradable polymers such as poly(lactic acid) (PLA), poly(glycolic acid) (PGA), and polycaprolactone (PCL). It has long been recognized

that hydroxyapatite (HA, $\text{Ca}_{10}(\text{PO}_4)_6(\text{OH})_2$) consists of about 60 wt% of bone¹³ in composite with collagen.¹⁴ Thus, mixing polymers with HA or other related calcium phosphate (*e.g.* α -TCP and β -TCP) to form micro/nano-composites has been a conventional strategy to improve the mechanical properties and bioactivity of scaffolds with considerable success both *in vitro* and *in vivo* for bone tissue engineering. Examples of degradable polymers that are composited with HA or related calcium phosphate include poly(L-lactic acid), poly(D,L-lactic acid), polycaprolactone, and poly(propylene fumarate) (PPF).¹⁵ However, seeking bioactive bone tissue engineering materials and scaffolds seems still to be an endless on-going effort and a challenge due to the unsatisfactory osteoinductivity and osteointegration and significant inflammatory responses of the existing bone biomaterials *in vivo*.

By carefully reviewing the literature on natural bone compositions, it is clear that one important fact has been inadvertently overlooked in the design of bone biomaterials and scaffolds in the past. Natural bone is an organic–inorganic nanocomposite where thin nanocrystals of apatite are embedded in collagen. The small nanocrystals (3 nm thick) control the mechanical properties of bone and prevent crack propagation. Early in the 1960s, it was found that citrate molecules consist of about 5 wt% of the organic components in bone, which was thought to regulate the stabilization of thin nanocrystal apatite.¹⁶ More recently, a careful solid-state

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nuclear magnetic resonance (NMR) study revealed that the surface of apatite nanocrystals is studded with strongly bound citrate molecules.¹⁷ Citrate is not a dissolved calcium-solubilizing agent but a strongly bound, integral part of the bone nanocomposite. Surprisingly, citrate was not even mentioned in most of the literature in the past 30 years, not even those on bone biomaterials and scaffolds.

The natural existence of citrate in bone hints that citrate should be considered in bone biomaterial and scaffold design although its function on bone development is still largely unknown. A recent study on poly(diols citrate)/HA composites further support this strategy. Poly(diols citrate) is the first citrate-based biodegradable elastomer that holds promise for tissue engineering and other biomedical applications.^{18,19} Use of poly (diols citrates) allows the formation of a degradable composite containing up to 60% of HA within it, while the conventional polylactide can only form a composite with up to 35% of HA. Poly(diols citrate)/HA demonstrated excellent osteoconductivity and induced rapid mineralization. Surprisingly, it elicited no chronic inflammatory response at the tissue-composite interface *in vivo*.¹⁹ It was inferred to that the rich -COOH from citrate units of poly(diols citrate) prompt calcium chelation thus facilitating poly(diols citrate)/HA interaction, enhancing mechanical properties, and promoting mineralization. Clear understanding on why poly(diols citrate)/HA exhibited excellent *in vivo* tissue/bone-compatibility is still unknown. Although poly(diols citrate)/HA was not developed under an hypothesis that the existence of citrate in bone may implicate the bone development, its excellent *in vivo* performance prompts further systematic study on the development of citrate-based orthopedic biomaterials.

With the goal of mimicking nature, it is our belief that citrate-based polymer/HA (or other related calcium phosphate) composite materials could constitute the next generation of bone tissue engineering and orthopedic biomaterials. Our lab has long been developing citrate-based biodegradable elastomers for versatile biomedical applications, including the recently developed crosslinked urethane-doped polyester (CUPE),^{20–22} poly(alkylene maleate citrate) (PAMC),^{23–25} and biodegradable photoluminescent polymer (BPLP).^{26,27} With careful molecular design, these citrate-based polymers offer versatile processability and tunable mechanical, degradable, and optical properties. These versatile citrate-based biodegradable polymers form a platform for us to understand and custom-design biomaterials toward bone tissue engineering and orthopedic applications.

Recently, significant attention has been placed on delivering various types of cells (autologous chondrocytes, mesenchymal stem cells, and osteoblasts) to treat irregularly shaped bone defects for bone regeneration. It has been recognized that *in situ* crosslinkable injectable hydrogels would be more beneficial for cell/drug delivery in filling irregular defects as they can homogeneously fit into the void *in situ* prior to setting *via* a minimally invasive injection. However, most of the synthetic hydrogels lack of osteoconductivity. Therefore, injectable composites composed of injectable hydrogels with

osteoconductive HA microparticles can be an alternative solution. Very recently, a few groups have started working on injectable hydrogel/HA composites for cell/drug delivery for orthopaedic applications using injectable polymers such as PPF,¹⁵ Poly(ethylene glycol)-diacrylate,²⁸ Poly(2-hydroxyethyl methacrylate),²⁹ poly(ethylene glycol)-poly(ϵ -caprolactone)-poly(ethylene glycol),³⁰ and acrylamide-based polymers.³¹ It has been confirmed that HA microparticles within the hydrogels not only enhance the mechanical strength but also serve as bioactive molecules to enhance cellular responses. However, none of these injectable composites considered the benefits of the aforementioned citrate effects in natural bone.

In the present study, a new bioactive citrate-based injectable biodegradable composite is developed for delivering cells to irregular bony defects. Given the aforementioned benefits of POC/HA used for orthopaedic applications, we have recently investigated and reported the synthesis and characterization of the first citrate-based water-soluble, *in situ* crosslinkable, and biodegradable polymers, poly (ethylene glycol maleate citrate) (PEGMC) for cell delivery.²⁴ These hydrogels were found to be elastic when compressed and demonstrated minimal cytotoxicity *in vitro* comparable to the FDA approved Poly (ethylene glycol) diacrylate (PEGDA) and completely degrade *in vivo* with minimal inflammatory responses.^{23,32} Thus, an injectable composite derived by combining PEGMC and HA particles would have characteristics potentially ideal for injectable orthopaedic applications. The objective of this study is to evaluate the physicochemical properties, polymer and HA interaction, cytotoxicity, and injectability of CPEGMC/HA composites potentially serving as injectable materials for osteoblast delivery in orthopaedic applications.

Experimental section

Materials

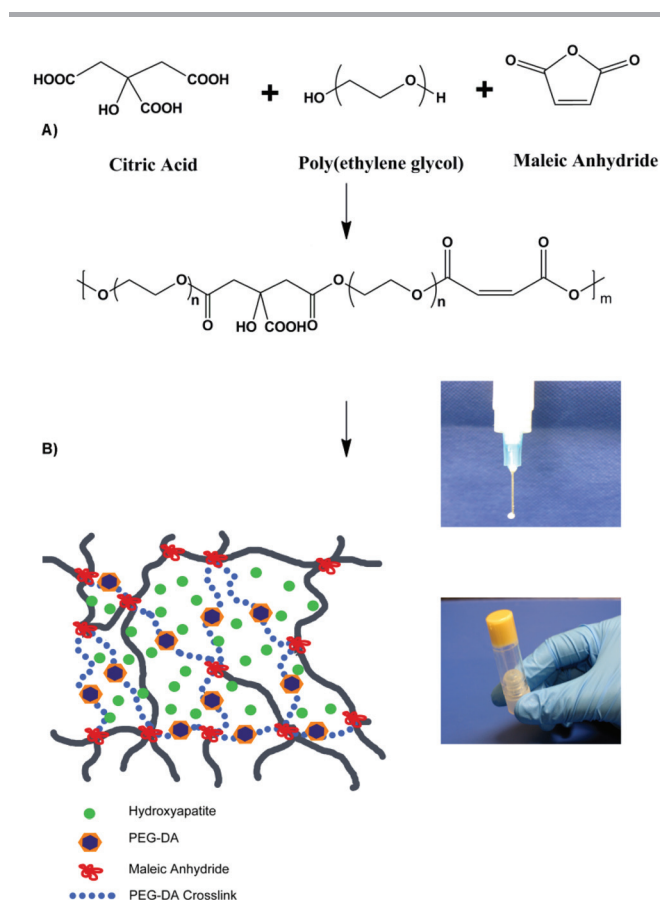
HA [M_w : 502.32, assay >90% (as $\text{Ca}_3(\text{PO}_4)_2$); particle size: >75 μm (0.5%), 45–75 μm (1.4%), <45 μm (98.1%)] was purchased from Fluka (St Louis, MO, USA). All other chemicals were purchased from Sigma Aldrich (St Louis, MO), except where mentioned otherwise. All chemicals were used as received.

Synthesis and preparation of CPEGMC/HA composite

PEGMC polymer was synthesized by a simple and controlled polycondensation reaction of poly(ethylene glycol) (PEG) ($M_w = 200$), citric acid (CA) and maleic anhydride (MA), as reported previously.^{23,24} In a typical procedure, PEG, CA and MA were added to a 250 ml round-bottom flask fitted with an inlet and an outlet adapter. The flask was immersed in a silicon oil bath at 160 °C at constant stirring under continuous flow of nitrogen. Following the melting of the monomers, the temperature of the oil bath was reduced to 140 °C until the desired PEGMC polymer viscosity was achieved. To remove any unreacted monomers and small molecular weight oligomers, PEGMC was dissolved in de-ionized (DI) water and dialyzed (Spectrum,

500 Dalton cut-off) for 2 days with frequent changes of DI water, followed by freeze-drying.

The HA content of the PEGMC/HA composite was defined by the weight percentage of HA over the entire composite. Purified PEGMC polymer was dissolved in DI water to make a 30% (w/v) solution in a container (centrifuge tube or culture dish). The pH of the mixture was adjusted to 5.5 by adding sodium bicarbonate under continuous monitoring of pH. HA powder was then added to the polymer solution under thoroughly mixing. In the next step, 15% (w/v) of poly(ethylene glycol diacrylate) (PEGDA) ($M_w = 3.4$ K) was added as a crosslinker followed by the addition of an aqueous radical initiator ammonium persulfate (APS) and N,N,N',N' -tetramethylethylenediamine (TEMED) at 25 mM concentration. The entire container was tightly sealed and then placed in an incubator at 37 °C for ~15 minutes to obtain crosslinked poly(ethylene glycol maleate citrate)/HA (CPEGMC/HA) composites. The schematic of PEGMC synthesis and CPEGMC/HA crosslinking is shown in Scheme 1. Three different compositions of CPEGMC/HA were fabricated where the weight ratios of PEGMC:HA was varied from 70:30, 55:45 and 40:60 to yield CPEGMC/HA (30), CPEGMC/HA (45) and CPEGMC/HA (60) respectively.



Scheme 1 (A) Schematic representation of PEGMC synthesis and crosslinking; (B) schematic representation of CPEGMC/HA composites. PEGMC is crosslinked using redox initiators APS/TEMED in the presence of HA and PEGDA to produce crosslinked PEGMC/HA network.

Characterization of CPEGMC/HA composites

The sol content, swelling ratio, and degradation of the CPEGMC/HA composites were investigated by a mass differential method. CPEGMC/HA composites with different HA composition were made as previously described. The CPEGMC/HA samples were cut into circular discs of dimensions 10 mm diameter by 2 mm thickness ($n = 8$). Samples were lyophilized for 48 hours to remove all water and weighed (M_i).

For the sol-gel fraction study, samples were immersed in 10 ml of 1,4-dioxane for 24 hours, freeze-dried and weighed (M_d). Sol fraction (SF) of the hydrogel composites was determined by using following eqn (1):

$$\%SF = \frac{M_i - M_d}{M_i} \times 100 \quad (1)$$

For the swelling study, the crosslinked composites were swollen in DI water to equilibrium swelling, blotted with Kimwipe paper, and weighed (M_w) and then lyophilized and weighed again (M_d). The swelling ratio (SR) was calculated using the formula in eqn (2)

$$\%SR = \frac{M_w - M_d}{M_d} \times 100 \quad (2)$$

For the degradation study, CPEGMC/HA disks were incubated in phosphate buffer saline (PBS) (pH = 7.4) at 37 °C. The buffer was changed regularly to ensure that the pH of the buffer was maintained at 7.4 during degradation. The initial weight of CPEGMC/HA discs was recorded as M_i . At pre-determined time points, the samples were removed from PBS, washed in DI water thrice to remove any residual salts and lyophilized and weighed (M_t). Degradation was determined by weight loss ratio as shown in eqn (3).

$$\%ML = \frac{M_i - M_t}{M_i} \times 100 \quad (3)$$

Microstructural characterization of CPEGMC/HA composites

The microstructures of the CPEGMC/HA composites were examined using a Leica DMLP microscope (Leica Microsystems Inc., Bannockburn, IL) fitted with a Nikon E500 camera (Nikon Corp., Japan). Hydrated CPEGMC/HA samples were first embedded in OCT freezing media (Polysciences Inc., Warrington, PA), cryosectioned into 10 μ m thick sections, and then photomicrographed to characterize the porous morphology of the composites and the distribution of HA in the composites.

Mechanical properties of CPEGMC/HA composites

Un-confined compressive mechanical tests of CPEGMC/HA composites were studied under a MTS Insight 2 mechanical tester equipped with a 10 N load cell (MTS, Eden Prairie, MN). Compressive tests of CPEGMC/HA at two different conditions were conducted, where CPEGMC/HA was compressed immediately after crosslinking (as prepared) and after equilibrium swelling (fully hydrated). Briefly, the cylindrical shaped specimens of dimensions 10 mm by 10 mm (diameter \times height)

were compressed at a rate of 1 mm min^{-1} to 50% strain. The initial modulus was calculated by measuring the gradient at 10% of compression of the stress–strain curve. Fully hydrated samples were subjected to cyclic compression to 20% strain for ten cycles and the hysteresis curves of load *vs.* strain were plotted. The results were presented as mean \pm standard deviation ($n = 5$). Cancellous bone cylinders removed from a femoral head of a domestic pig were decellularized and subjected to compression tests as a control.

***In vitro* mineralization on CPEGMC/HA composites**

Simulated body fluids (SBF) were prepared as described previously^{33–35} to study *in vitro* mineralization on the CPEGMC/HA composites. The SBF with an inorganic ion concentration five times human blood plasma was denoted as SBF (5 \times). Disk shaped polymer scaffolds were immersed in 6-well plates containing SBF (5 \times) for up to 7 days. The SBF was replaced every other day. After incubation for various periods of times, the specimens were washed carefully with DI water to remove any soluble inorganic ions, and dried in air and coated with silver prior to analysis under a Hitachi 3000N scanning electron microscope (SEM) (Hitachi, Pleasanton, CA).

***In vitro* cell culture on CPEGMC/HA composites**

CPEGMC and CPEGMC/HA films were cut into cylindrical discs (10 mm in diameter and 1 mm in thickness). The samples were sterilized by soaking in 70% ethanol for 30 minutes, then exposing to UV light for 1 hour, and then washed with PBS for 3×5 minutes. Human fetal osteoblasts (hFOBs) (ATCC) were cultured according to ATCC protocol in phenol free Dulbecco's modified Eagle's medium (DMEM) – Ham's F12 1 : 1 medium supplemented with 10% fetal bovine serum (FBS) (HyClone) and geneticin ($300 \mu\text{g ml}^{-1}$; Sigma-Aldrich, St. Louise). The culture flasks were kept in an incubator maintained at $37 \text{ }^\circ\text{C}$, 5% CO_2 , and 95% relative humidity. The cells were trypsinized, centrifuged, and suspended in media to obtain a seeding density of 2×10^5 cells per ml on CPEGMC/HA films. After 48 hours of culture, the cells were fixed in 3% (v/v) of glutaraldehyde in PBS and sequentially dehydrated with a graded series of ethanol, lyophilized, and sputter coated with silver. The samples were then observed under SEM to view the morphology of the attached cells. In addition to SEM images, films seeded with hFOBs were stained with carboxyfluorescein diacetate, succinimidyl ester (CFDA-SE) (Invitrogen, Carlsbad, CA) green fluorescent cell tracer according to the manufacturer's protocol and viewed under a Zeiss Axiovert inverted microscope (Carl Zeiss Micro-Imaging, Thornwood, NY).

The cytotoxicity of the composites was further evaluated by encapsulating hFOB cells within the networks of the CPEGMC/HA composites. Briefly, PEGMC solution was formulated as described in Section 2.2. The solution was then sterilized *via* filtering through $0.22 \mu\text{m}$ filter. hFOBs were added to polymer solution to achieve a final cell density of 15×10^6 cells per ml. After being held at $37 \text{ }^\circ\text{C}$ for approximately 15 minutes in an incubator, the cells/composite constructs were transferred to a

6 well plate and incubated in osteoblast media. During the 21-day culture period, the culture media was changed every 48 hours and the cell-encapsulated composites were taken out at week 1, 2 and 3 and stained using LIVE/DEAD assay (LIVE/DEAD Viability/Cytotoxicity Kit, Invitrogen, Carlsbad, CA). Cell viability and distribution were observed under a Zeiss Axiovert inverted microscope (Carl Zeiss MicroImaging, Thornwood, NY).

For biochemical analysis, at various time points, the cell-cultured CPEGMC/HA constructs were removed from media and homogenized in PBS. $500 \mu\text{l}$ of 0.2% Triton X-100 solution was added to the constructs. The samples were subjected to two freeze-thaw cycles. The homogenates were then sonicated for 30 s. The homogenates were centrifuged for 10 minutes at 4000 rpm. The supernatant after centrifuging was used to measure the DNA content of each construct using PicoGreen assay according to the manufacturer protocol. Measurements were performed in triplicate.^{36,37}

To characterize the alkaline phosphatase (ALP) activity within the composite, alkaline buffer solution was added to the constructs and sonicated to ensure that the constructs were completely homogenized. The lysate was centrifuged at 4000 rpm for 10 minutes at $4 \text{ }^\circ\text{C}$. $100 \mu\text{l}$ of supernatant was incubated with $100 \mu\text{l}$ of *p*-nitrophenyl phosphate solution in a 96 well plate at $37 \text{ }^\circ\text{C}$ for 1 hour. The reaction was stopped by adding 0.2 M NaOH solution to each well. The production of *p*-nitrophenyl in the presence of ALP was measured by monitoring the light absorbance at 405 nm using a microplate reader (Infinite 200, Tecan Group Ltd, Switzerland).³⁸

To quantify the calcium content within the composite, supernatant was collected and $50 \mu\text{l}$ of the supernatant was added to a 96 well plate. The calcium concentration of the cell lysate was analyzed using cresolphthalein complexone (Sigma). After 10 minutes, the absorbance was read at 575 nm using the Tecan microplate reader (Infinite 200).³⁸

***Ex vivo* injectability of the CPEGMC/HA composite**

The *ex vivo* study was conducted on a porcine femoral head. Femoral head was cored to create a cylindrical cavity. PEGMC/HA solution was injected into the cavity and incubated at $37 \text{ }^\circ\text{C}$ for *in situ* crosslinking. To visualize the *in situ* formed composites, the femur head was further sectioned and photographed using a digital camera.

Statistical analysis

Data was expressed as mean \pm standard deviation. The statistical significance between two sets of data was calculated using one-way ANOVA. Data were taken to be significant, when $p < 0.05$ was obtained.

Results

Polymer synthesis, cross linking and characterization of PEGMC/HA composite

PEGMC polymer was synthesized by a simple, one pot, and controlled polycondensation reaction from maleic anhydride,

PEG, and citric acid. The polymer chains consist of degradable ester bonds, crosslinkable carbon-carbon double bonds, and pendant carboxyl and hydroxyl functional group as confirmed by previous FTIR and $^1\text{H-NMR}$.²³ Due to these pendent carboxylic groups present throughout the polymeric network, there are strong interactions between the calcium ions of HA molecules. This was confirmed in Fig. 1, as the peak maxima due to bond vibration of hydroxyl groups at 3570 cm^{-1} of the polymer shift to 3590 cm^{-1} as in the case of PEGMC/HA composite where the characteristic HA peaks (1049 cm^{-1} for PO_4^-) and degradable ester bonds (C=O at $1690\text{--}1750\text{ cm}^{-1}$ and C-O at $1000\text{--}1260\text{ cm}^{-1}$) are still evident at their respective wave numbers. PEGMC is completely soluble in water and can be homogeneously mixed with inorganic minerals such as hydroxyapatite without observable sedimentation before, during, and after the crosslinking process. In our previous study, a detailed rheology study of the PEGMC/HA composite demonstrated that the hydrogel composite precursor is highly injectable (*via* a 27-gauge needle) with low viscosity even in higher (50%) HA content and can be crosslinked through radical polymerization aided by PEGDA crosslinkers within 3 to 15 minutes at $37\text{ }^\circ\text{C}$ to create mechanically stable hydrogel composites.³²

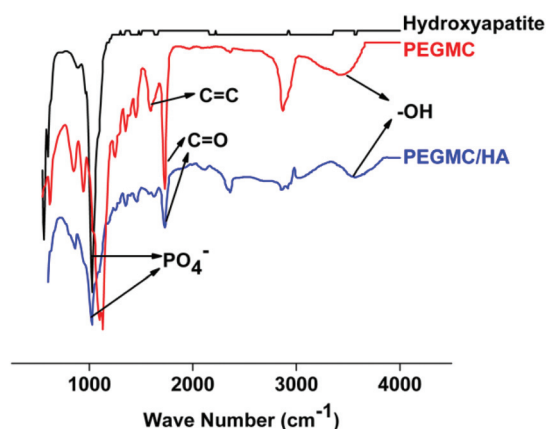


Fig. 1 FTIR spectra of HA powder, PEGMC hydrogel and CPEGMC/HA composite.

Fig. 2A showed the overall percentage of leachable (sol) content of the crosslinked composites. The sol content did not vary significantly with increasing HA concentration. For example, The sol content for CPEGMC/HA(30), CPEGMC/HA(45), and CPEGMC/HA(60) were $6.90 \pm 0.74\%$, $7.06 \pm .28\%$, and $5.66 \pm 1.13\%$, respectively. Fig. 2B showed the swelling kinetics of CPEGMC/HA containing various concentration of HA particles. These hydrogel composites started uptaking water and reached equilibrium swelling as early as 10 minutes and did not significantly increase swelling till 12 hours. Increasing the amount of HA in composites decreased swelling ratios. For instance, the swelling ratio reduced from 9.56 ± 0.56 (CPEGMC/HA(30)) to 7.89 ± 0.96 (CPEGMC/HA(45)) and 3.42 ± 0.66 (CPEGMC/HA(60)).

The degradation profile of CPEGMC/HA in PBS showed increasing mass losses with lowering concentrations of HA. As shown in Fig. 2C, CPEGMC/HA (30) showed a mass loss of $58.18 \pm 6.60\%$ at the end of 22 weeks. The high HA concentration CPEGMC/HA(60) showed a mass loss of $32.24 \pm 1.79\%$ at the end of 22 weeks. This result indicated that the mass loss of the composites was primarily due to the degradation of the polymers and that HA particles did not leach out from the network in noticeable amounts during degradation. The composites maintained their integrity throughout the 22 weeks of degradation study.

Microstructure characterization of CPEGMC/HA composites

The pore morphology of the fully hydrated CPEGMC/HA composite sections ($5\text{ }\mu\text{m}$) was observed under microscope. It was observed that the pure PEGMC gel network was highly porous with a pore size of $200\text{--}400\text{ }\mu\text{m}$ (Fig. 3A). Multiple sections of the fully hydrated cylindrical hydrogels cut at various locations and angles were analyzed and observed with similar morphologies. This technique allowed us to visualize the interaction of the polymer and HA particles under fully hydrated conditions which is otherwise impossible to observe by SEM or cryo-SEM.²⁸ The pore structure of the composites was not altered with the addition of HA particles. In fact while under a fully hydrated state, these mineral particles were distributed within the polymer matrix rather than simply encapsulated in

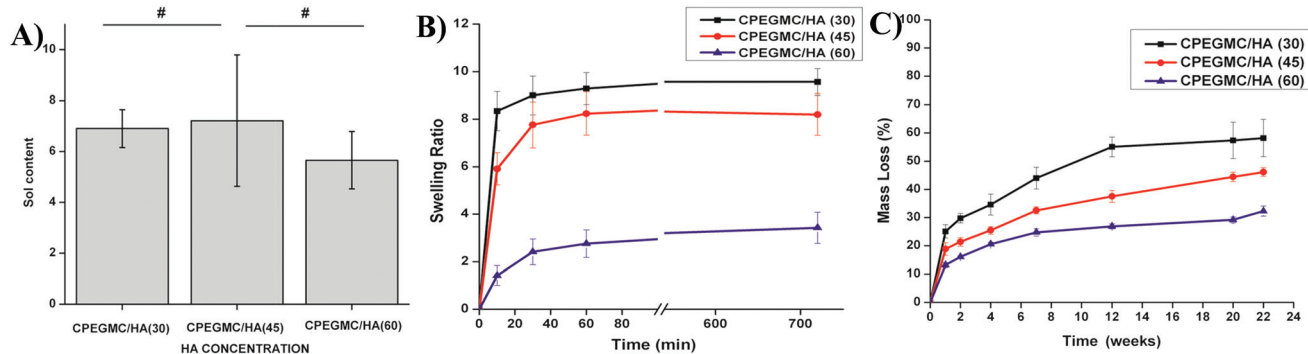


Fig. 2 Physiochemical characterization of CPEGMC/HA. (A) Degree of sol content as a function of HA concentration; (B) degree of swelling as a function of HA concentrations; (C) *in vitro* mass loss of CPEGMC/HA in PBS (pH 7.4; $37\text{ }^\circ\text{C}$) over time up to 22 weeks.

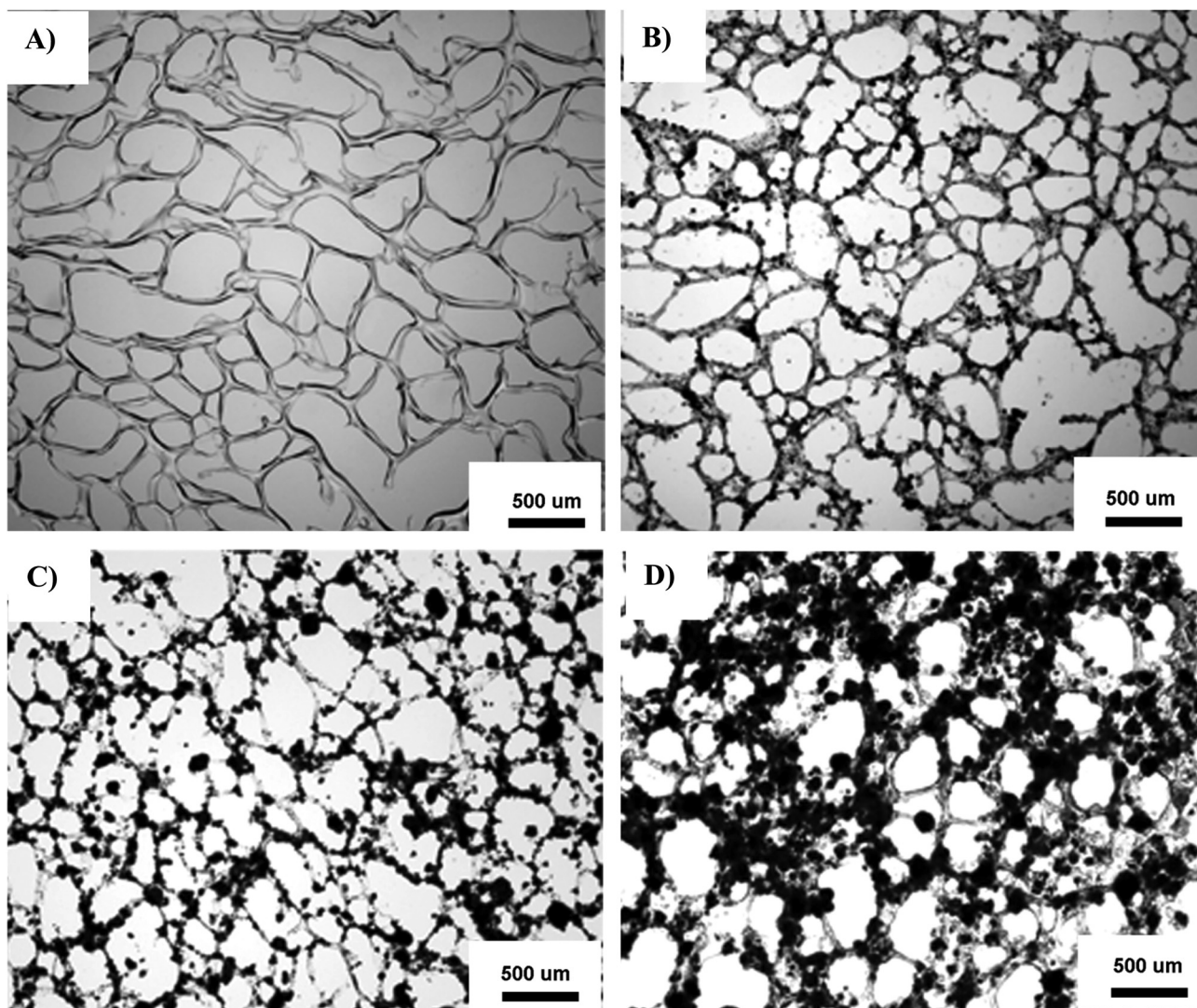


Fig. 3 Representative light microphotographs of macroporous CPEGMC hydrogel and CPEGMC/HA composite thin sections (10 μm). (A) CPEGMC; (B) CPEGMC/HA (30) composite; (C) CPEGMC/HA(45) composite; and (D) CPEGMC/HA(60) composite. The black dots denote the HA microparticles within the polymer matrix. The white spaces represent void pore spaces.

the pores. As shown in Fig. 3B–D, the more HA was incorporated in the composite, the more HA could be observed within the polymer matrix.

Mechanical properties of CPEGMC/HA composite

CPEGMC/HA composites with varying HA content (30, 45, 60)% were prepared and subjected to unconfined compressive tests as soon as they were freshly prepared (as-prepared) and after equilibrium swelling (fully hydrated). As shown in Fig. 4A–B, the modulus of the as-prepared CPEGMC/HA(30) (10 ± 5 kPa) did not change significantly compared to CPEGMC/HA(45) (26 ± 12 kPa). The modulus of CPEGMC/HA(30) (12.5 ± 5 kPa) and CPEGMC/HA(45) (30 ± 7 kPa) also showed no significant difference when tested after full hydration. However, the modulus of CPEGMC/HA(60) (205 ± 26 kPa) was significantly increased compared to CPEGMC/HA(30) and CPEGMC/HA(45) and it also significantly decreased in the fully hydrated state

(100 ± 20 kPa). Good overlaps of the stress–strain curves were observed when 10 consecutive compressive loading and unloading cycles with 20% strain on fully hydrated CPEGMC/HA samples (Fig. 4C). These mechanical results demonstrated that the CPEGMC/HA confers the constructs with excellent elastic properties and the HA contributed to the increased strength. As shown in Fig. 4D, increasing the crosslinking times also increases the composite mechanical strength. The dried CPEGMC/HA (60) demonstrated a close ultimate compressive strength to that of decellularized cancellous bone of femur head of porcine model.

In vitro mineralization

When incubated in simulated body fluid, by day 1, homogeneous nucleation and 2-dimension growth of calcium phosphate was evident as mineral nodules appeared on the surface of CPEGMC/HA composite (Fig. 5A). By allowing

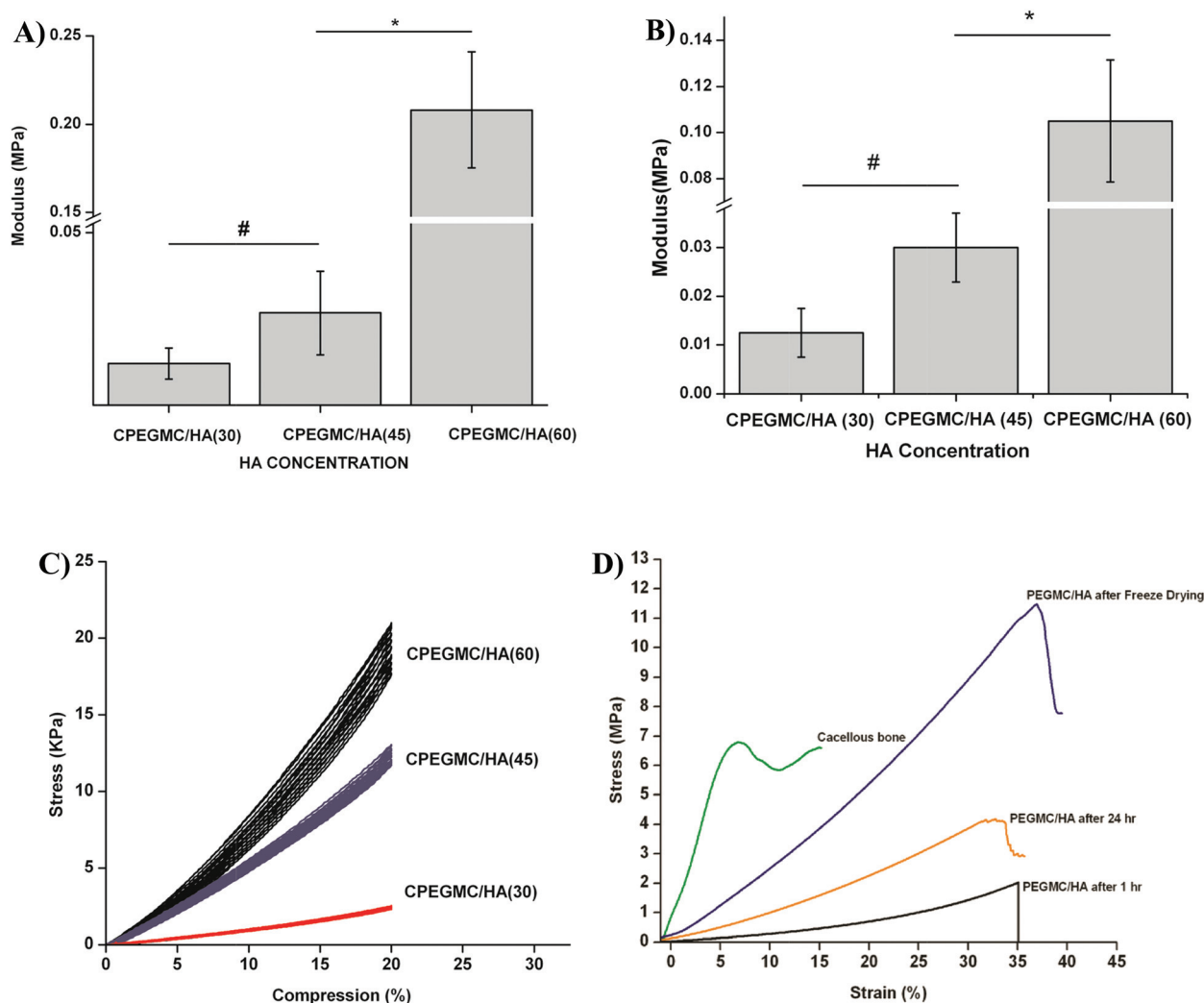


Fig. 4 Mechanical tests of as-prepared and fully hydrated CPEGMC/HA composites. (A) Compressive modulus of as-prepared CPEGMC/HA composites; (B) compressive modulus of fully hydrated CPEGMC/HA composites; (C) cyclic compression tests of fully hydrated CPEGMC/HA composites. Ten consecutive cycles of controlled loading and unloading was applied to fully hydrated CPEGMC/HA specimens using MTS Insight II mechanical tester using a 10 N load cell. (D) Compression test of decellularized cancellous bone and synthetic hydrogel composites at different conditions.

mineralization to proceed for a longer period of time (day 7), more distinct mineral nodules continuously covered the entire surface (several micron) of the CPEGMC/HA composite (Fig. 5B). Upon evaluation by the calibrated energy dispersive spectroscopy (EDS), the apatite nodules on the surface of the composite revealed a Ca/P ratio (1.6 ± 0.1) similar to that of synthetic HA.

In vitro cell culture

hFOB seeded on the surface of the PEGMC/HA composites supports cell adhesion and promotes cell spreading throughout the surface over the experimental period of 48 hours when observed under fluorescent microscope and SEM (Fig. 6A and B).

hFOB cells were further encapsulated within the CPEGMC/HA hydrogel composites to test their potential for cell encapsulation/delivery. It was found that the cells survived the encapsulation procedure and proliferated during the 3 weeks of

subculture. Fluorescent images of the encapsulated osteoblasts stained by LIVE/DEAD assay at day 1 and day 21 of encapsulation are shown in Fig. 6C and D, where red fluorescence denotes dead cells and green fluorescence indicates live cells. In Fig. 6D, the majority of hFOB cells were green, indicating that the CPEGMC/HA polymer network provided a cytocompatible environment for hFOB delivery. Only a few cells died under the crosslinking conditions. Furthermore, the images also indicated a uniform cell distribution in the composites. hFOBs could adhere and spread on CPEGMC/HA composites, which could be clearly seen in Fig. 6E.

The deoxyribonucleic acid (DNA) content of hydrogels containing osteoblasts was assessed using the PicoGreen assay and the results were shown in Fig. 6F. DNA content increased from week 1 to week 3, demonstrating that cells were proliferating. No significant difference was found from both the pure PEGMC group and the CPEGMC/HA group at each time point.

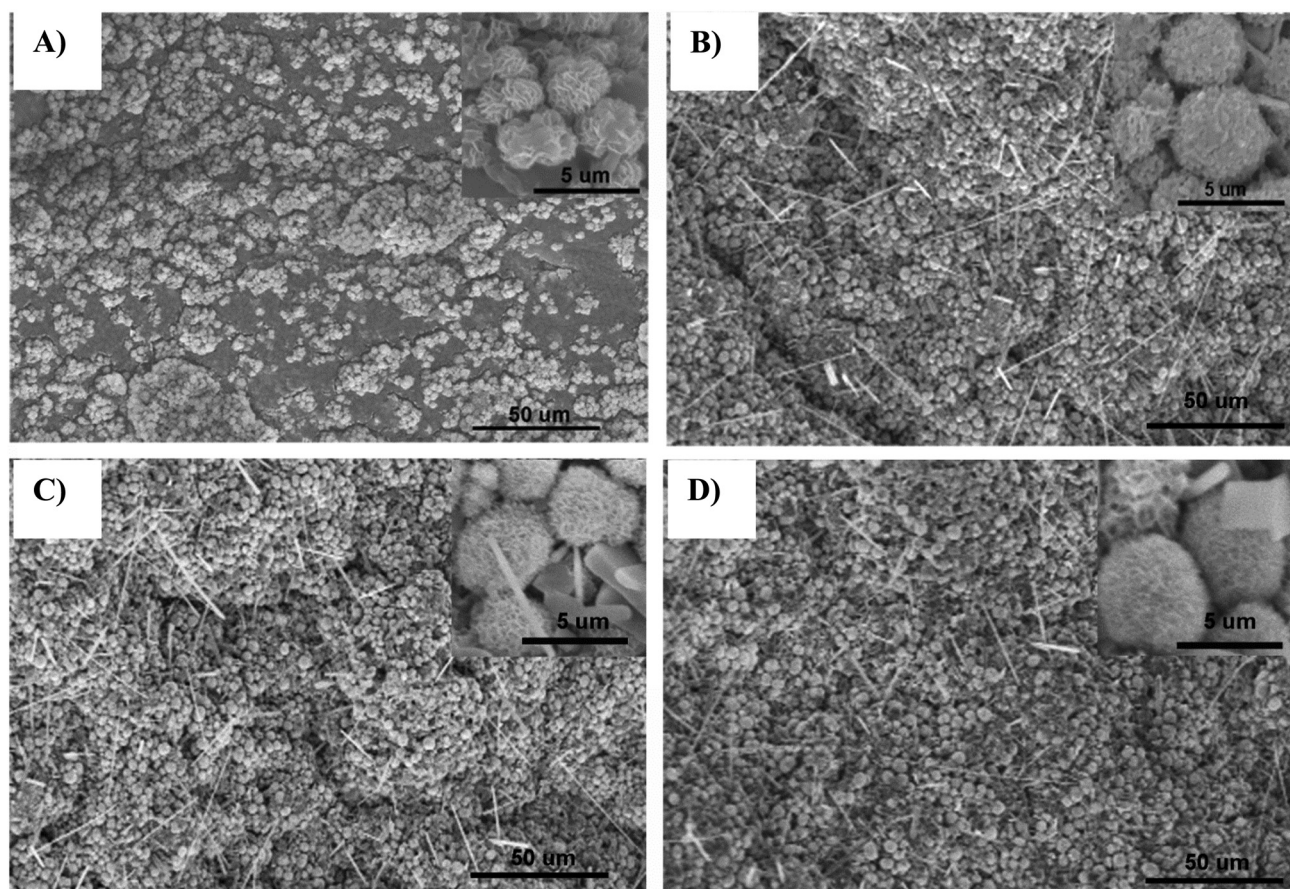


Fig. 5 Scanning electron microscope images of *in vitro* mineralization of CPEGMC/HA composite. (A) CPEGMC/HA (45) composite when incubated in 5x SBF for 1 day; (B) CPEGMC/HA(45), (C) CPEGMC/HA(30), and (D) CPEGMC/HA(60) composites when incubated in 5x SBF for 7 day.

As shown in Fig. 6G, the ALP production of the encapsulated hFOBs in CPEGMC/HA showed a significant increase after two weeks. In contrast, the ALP production on pure CPEGMC did not show any significant change within 3 weeks of culture. The calcium content deposited by the hFOBs in CPEGMC/HA was significantly higher compared to that of the pure CPEGMC and CPEGMC/HA composites without cells. The significant increase was observed at the 2nd week (Fig. 6H).

Ex vivo study

An *ex vivo in situ* delivery study was conducted on a porcine femoral head (Fig. 7A–D). CPEGMC/HA could be easily injected into the defect in the central region of the femoral head using a syringe fitted with a biopsy cannula. No leakage was observed prior to crosslinking. Visual examination of the polymerization site after sectioning the femoral head suggested that the *in situ* crosslinked CPEGMC/HA completely filled up the irregular implantation site and reinforced the femoral head.

Discussion

Hydroxyapatite (HA, $\text{Ca}_{10}(\text{PO}_4)_6(\text{OH})_2$) constitutes around 60 wt % of bone and has long been recognized as a crucial

biomaterial to design tissue-engineered bone substitutes.¹³ It has been confirmed that HA and related calcium phosphate (e.g. α -TCP and β -TCP) not only enhances the mechanical properties but also plays a critical role in the osteoconduction and osseointegration of the implanted bone graft.^{39,40} Thus, developing composite materials based on biodegradable polymers and HA became an intense focus in bone tissue engineering. Such composites take advantage of the formability of polymers and the bioactivity of HA to enhance the mechanical properties of the fabricated scaffolds. In the meantime, the poor bioactivity of most of synthetic polymers can be improved.⁴¹ Examples of degradable polymers compositing with HA or related calcium phosphate include poly(L-lactic acid), poly(D,L-lactic acid), polycaprolactone, and poly(propylene fumarate).¹⁵

These prefabricated polymer composites may be an excellent choice for critical bone defects, however, may not be applicable in the case of irregular bone defects, as in the case of femoral head osteonecrosis where complete evacuation of the necrotic tissue is performed by core decompression. It is desirable to use an injectable scaffolding system that can perfectly fill the void, solidifies within the site of injection, and induce/promote bone tissue regeneration. In fact, hydrogels based upon materials such as gelatin,⁴² poly(propylene fumarate),¹⁵ polymethylmethacrylate,⁴³ and hydroxypropylmethyl cellulose⁴⁴

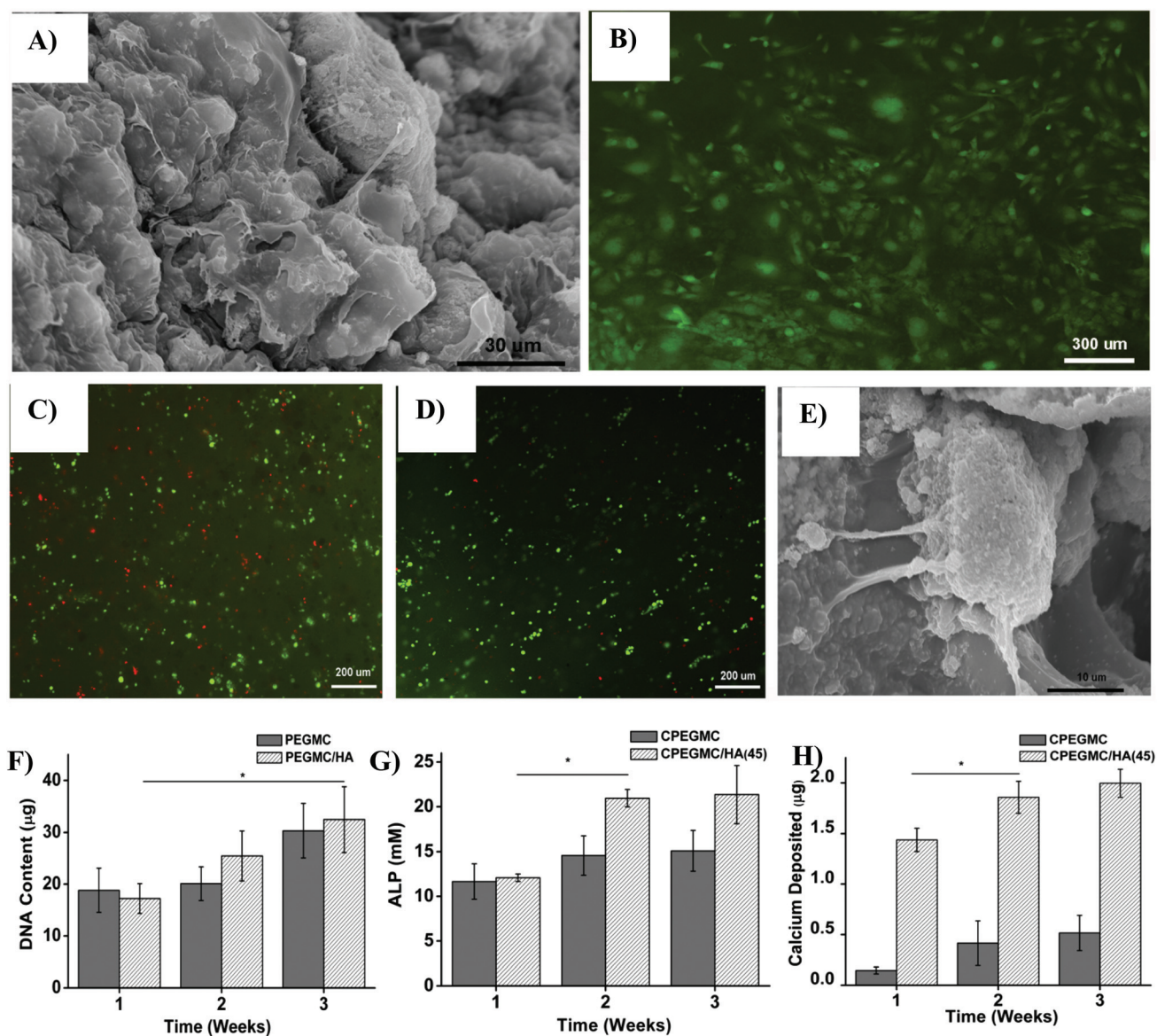


Fig. 6 hFOBs attachment on the surface of CPEGMC/HA (45) films 48 hours post-seeding. (A) SEM image of hFOB-seeded PEGMC/HA(45) composite and (B) micro-photograph of hFOBs on CPEGMC/HA(45) composite stained with CFDA-SE fluorescent dye. hFOBs encapsulated in CPEGMC/HA(45) composites. (C) Live/Dead stain of hFOBs encapsulated in CPEGMC/HA(45) after 1 day culture; (D) live/dead stain of hFOBs encapsulated in CPEGMC/HA(45) after 21 days culture; live cells are fluorescent green and dead cells are fluorescent red; (E) cross-sectional SEM image of hFOBs encapsulated within the matrix; (F) DNA content assay; (G) alkaline phosphatase (ALP) assay; and (H) calcium assay. In F–H, hFOBs were encapsulated in CPEGMC and CPEGMC/HA(45).

have been combined with HA particles as a non-invasive injectable composite system for bone tissue replacement.

Herein, we reported a new elastomeric injectable PEGMC/HA composite for bone cell delivery. The water soluble PEGMCs would be a good choice of biomaterial for injectable application, as it has demonstrated excellent injectability, *in situ* crosslinkability, the presence of pendant functional groups, elastic mechanical properties, controlled degradability, and excellent cytocompatibility both *in vitro* and *in vivo*.

In a previous study, we have systematically studied the viscoelastic properties of newly developed citric acid-based injectable biodegradable PEGMC/HA composites *via* rheology

study. It was demonstrated that the intrinsic affinity of the pendent carboxyl group in the bulk of crosslinked PEGMC (CPEGMC) matrix to the HA particles provide a strong interfaces between the organic and inorganic components, a mimic to the natural citrate-bound apatite crystals in bone. The dispersion of HA particles within the matrix of PEGMC have very little effect on the gelation kinetics of chemically crosslinked gels but help to improve the elastic moduli due to ionic interactions between HA molecules and pendant $-\text{COOH}$ groups from the citrate units of the polymer.³² Consequently, this interaction provides even dispersion of HA particle throughout the polymer matrix and further avoids rapid material degradation.

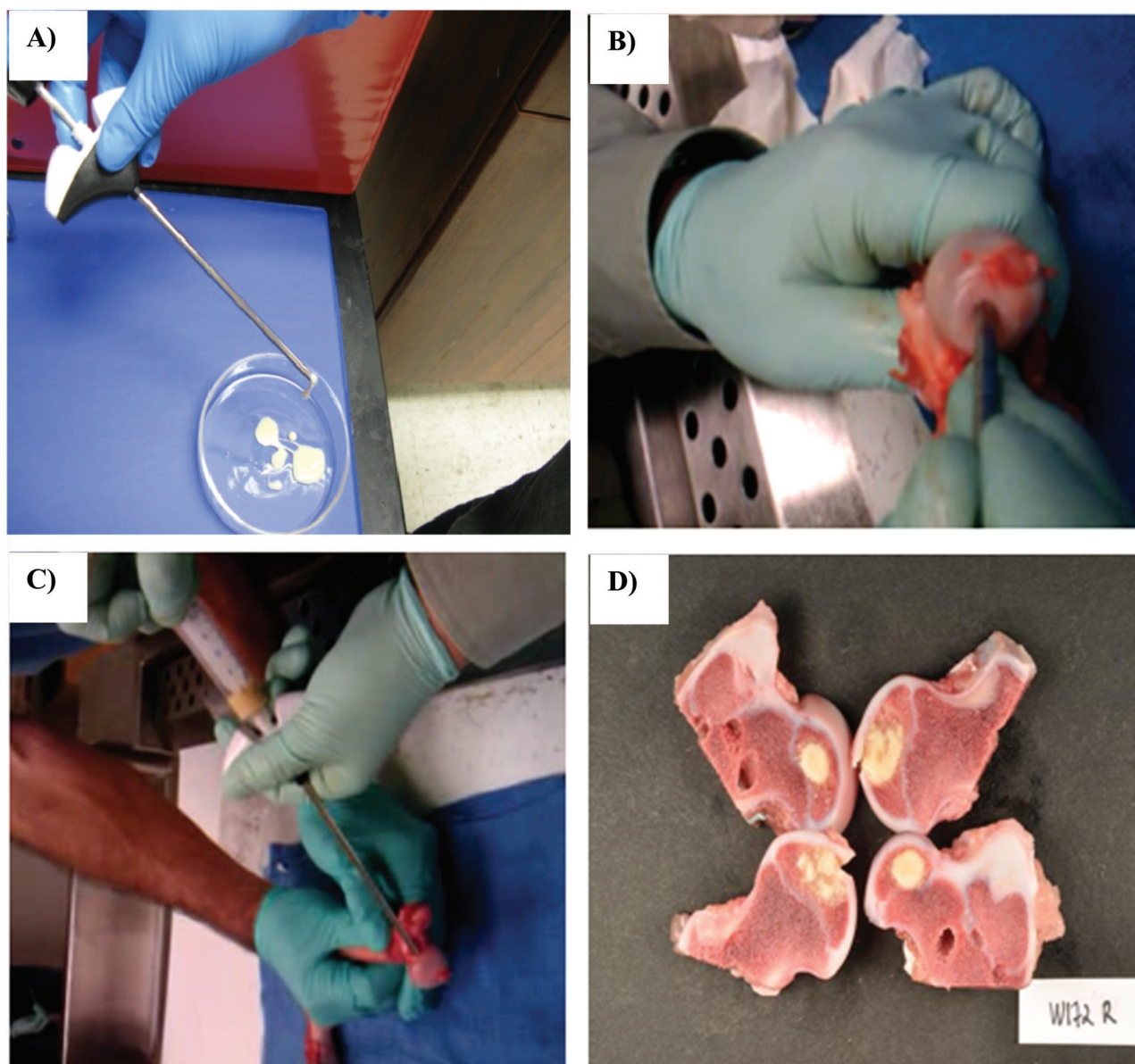


Fig. 7 *Ex vivo* study demonstrates that biodegradable injectable PEGMC/HA(45) can be injected into collapsed femoral head for reinforcement. (A) PEGMC/HA(45) solution loaded in cannula injection tool; (B) The cored femoral head with a cylindrical cavity; (C) PEGMC/HA(45) being injected into collapsed femoral head; and (D) the sectional view of cemented femoral head with CPEGMC/HA(45) composite. Crosslinking was achieved within 5 minutes of injection.

One of the prime concern regarding injectables is the leachable (sol content) component following the solidification. Especially for hydrogel/particle composites, the likelihood of particle leaching in a fully hydrated condition is high. As demonstrated earlier, PEGMCs were found to have a lower sol content after crosslinking (<15%, w/w).²³ Similar results were found with our CPEGMC/HA composites. In fact, an increase in the HA component did not show any significant increase in leachable components from the system. HA particles were also found to be homogeneously dispersed within the matrix without any settling by gravity during solidification. Excellent structural integration of the organic and inorganic

components were evident through FTIR spectroscopy (Fig. 1) and morphology analysis (Fig. 2). Furthermore, no evident amount of HA particles leached out from the composite for 22 week of incubation in PBS during the degradation study. Another crucial parameter to be addressed while designing bone scaffolds is its porosity.⁴⁵ Cancellous bone, where osteonecrosis occurs, is a highly porous environment with 50–90% porosity with average pore size of 10–400 μm to facilitate nutrient exchange and osteoblast cellular infusion.⁴⁶ Interestingly, CPEGMC/HA composites were found to be highly porous with open and interconnected pores within the range of 200 to 400 μm as similar to other hydrogel systems. An increase in the HA

content did not influence pore formation within the formulations that were investigated. Rather, the particles distributed evenly within the polymer matrices.

In our previous study, redox crosslinked PEGMC hydrogel degraded almost 70% of its mass by the end of one month in PBS at 37 °C.²³ Interestingly, even at the end of 22 weeks of an *in vitro* degradation study, the CPEGMC/HA composites were found to maintain their structural integrity without any noticeable amount of HA release from the system. It was inferred that the residual polymer network could still well chelate with HA particles to maintain the integrity of the composites. The morphology change of the composites at longer degradation times needs to be studied in the future. Nonetheless, CPEGMC/HA maintained excellent integrity during degradation, a favourable factor when used for tissue engineering applications. The degradation of CPEGMC/HA networks *in vivo* will be reported in our future studies as others have suggested that local enzymes and biological environments may impact material degradation *in vivo*.^{47–52}

In general, polymers themselves have poor osteoconductivity. Therefore, the incorporation of HA into polymers to induce mineralization has been a common way in designing bone tissue engineering composite biomaterials.^{53,54} It was reported that the apatite formed as a result of biomimetic process could have biological properties that are more similar to natural bone mineral than that of synthetic calcium phosphate bioceramics and provide a more conducive environment for bone formation^{38,55,56} and differentiation of mesenchymal cells to osteoblasts. Studies have shown that osteoblasts seeded on PLGA/HA scaffolds covered with a bone like apatite coating boost cell growth, ALP activity, and mineralization. Thus, bone like apatite may provide a favourable environment for osteoblast attachment and growth.⁵⁷ Inducing mineralization is one important property that should be inherited by bone scaffolds. Various strategies have been employed in biomaterials to demonstrate template-driven biomineralization of the system. For example, urea mediated surface hydrolysis was performed on poly(2-hydroxyethyl methacrylate) (pHEMA) to create a pendent carboxylic group with an aim to create high mineral content hydrogel.¹⁴ Our CPEGMC/HA is saturated with carboxylic acid groups from citrate molecules of CEPGMC that not only provide strong HA chelation within the network but also promote high affinity nucleation and growth of calcium phosphate in simulated body fluid. After 7 days of incubation, SEM micrographs indicated a robust surface mineral layer due to apatite deposition covering entire surface of the composite. The induced apatite crystals on the composite hydrogel demonstrated typical ‘cauliflower’. This data clearly demonstrates that the PEGMC/HA composite is a promising candidate for bone tissue engineering with high affinity for calcium ions.

Material stiffness has been reported to play an important role in adhesion, proliferation, and differentiation of cells seeded/encapsulated on/in biomaterials. Studies showed that a matrix with a stiffness of 20–110 kPa could promote differentiation of mesenchymal stem cells (MSCs) into osteoblasts.⁵⁸ Thus, we formulated hydrogel composites mimicking similar

stiffness. The as-prepared CPEGMC/HA composites exhibited stiffness within 110 kPa. Given the results, our composites injected into bone defects should exhibit favourable mechanical stiffness conducive for bone differentiation. It is also noteworthy that complete hydration did not sacrifice the ability of CPEGMC/HA composites to withstand cyclic compression with 20% compressive strain. Furthermore, the mechanical properties of CPEGMC/HA composites can be further tuned by adjusting the polymer concentrations and crosslinking times.

In addition to surface mineralization, PEGMC/HA could also serve as a template for surface cell adhesion and proliferation. Upon implantation, it is extremely important for scaffolds to encourage cell adhesion and tissue infiltration within the scaffold as it helps to improve tissue-material integration. Furthermore, we encapsulated osteoblasts within the matrix of PEGMC/HA composite and evaluated their functionalities for a 3 week period. Although minimal cytotoxicity was expected during the process of encapsulation, viable cells within CEPGMC/HA proliferated and displayed characteristic osteoblastic functionality in bone formation. ALP, an ectoenzyme usually produced in the early stages of osteoblastic differentiation,⁵⁹ was found to be significantly increased by week 2, in agreement with the increase of calcium deposition at week 2 and week 3 of the subculture. In a review of the literature, it was rarely reported that cells could be encapsulated in injectable polymer/HA composites. Most of the cell encapsulation carriers are pure hydrogels with or without bioconjugation of biological molecules.^{60,61} The presence of HA within injectable PEGMC/HA composites favoured ALP production and calcium deposition as compared to CPEGMC gels alone.

To demonstrate the feasibility of injectable PEGMC/HA, an *ex-vivo* study was conducted on a porcine femoral head after the removal of the cancellous bone from its central region. The experiment clearly showed that the composite precursor can be easily injected and crosslinked *in situ* after completely filling the entire void space of the collapsed femoral head prior to crosslinking. PEGMC/HA with a HA concentration up to 60% was still injectable and crosslinkable *in situ*. Once fully filled with PEGMC/HA, the pocketed femoral head could be reinforced without collapse.

Conclusion

In summary, an injectable citrate-based PEGMC/HA hydrogel composite was developed. PEGMC/HA demonstrated highly porous micro-architecture, tunable mechanical and degradable properties, and exhibited excellent osteoconductivity supported by increased ALP production and calcium deposition by the cells seeded/encapsulated on/into the composites. PEGMC/HA with high concentrations of HA (up to 60%) could be injected into porcine femoral head bone defects and reinforce it within a short crosslinking time. By mimicking the citrate-bound apatite crystals, PEGMC/HA was the first citrate-based injectable composite scaffold with great potential serving as a cell delivery carrier for bone tissue engineering.

The discovery of citrate-bound apatite crystal in natural bone left many unanswered questions.¹⁷ For example, where is the citrate formed in bones? When does the citrate come into play for bone development? How does the abundance of citrate on the crystal surface influence on bone formation? These questions can be translated into questions in bone biomaterial design. How much citrate should be placed in bone biomaterials? How long and when should the citrate exist in biomaterials? Although our current study does not answer the above questions, it constitutes an initial step in the development of a citrate-based injectable biomaterial platform to study the above questions. The development of injectable PEGMC/HA also expands the repertoire of existing orthopaedic biomaterials.

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